

This article was downloaded by:

On: 27 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713618290>

Synthesis and Multinuclear NMR Study of a New Benzoyl Methylene Triparatolylphosphorine Ylide and its Reactions with Mercury (II) Halides

Seyyed Javad Sabounchei^a; Sepideh Samiee^a; Kazem Karami^a

^a Chemistry Department, Bu-Ali-Sina University, Hamadan, Iran

To cite this Article Sabounchei, Seyyed Javad , Samiee, Sepideh and Karami, Kazem(2006) 'Synthesis and Multinuclear NMR Study of a New Benzoyl Methylene Triparatolylphosphorine Ylide and its Reactions with Mercury (II) Halides', *Phosphorus, Sulfur, and Silicon and the Related Elements*, 181: 2, 447 — 452

To link to this Article: DOI: 10.1080/104265091001290

URL: <http://dx.doi.org/10.1080/104265091001290>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Synthesis and Multinuclear NMR Study of a New Benzoyl Methylene Triparatolylphosphorine Ylide and its Reactions with Mercury (II) Halides

Seyyed Javad Sabounchei

Sepideh Samiee

Kazem Karami

Chemistry Department, Bu-Ali-Sina University, Hamadan, Iran

The reaction of bromoacetophenone with triparatolylphosphine in chloroform solution gives [Hooch (p-tolyl)₃](I); then reaction of this new ylide with mercury (II) halides (HgCl₂, HgBr₂, Hgl₂) in equimolar ratio leads to pure compounds of [(p-tolyl)₃PCHCOC₆H₅][HgCl₂]₂ (II), [(p-tolyl)₃PCHCOC₆H₅]. HgBr₂]₂ (III), and [(p-tolyl)₃PCHCOC₆H₅][Hgl₂]₂ (IV) complexes. The IR ¹H¹³C, and ³¹P NMR together, with microanalysis data of the products, were obtained.

Keywords Mercury (II) halides; phosphorus ylide; triparatolylphosphine

INTRODUCTION

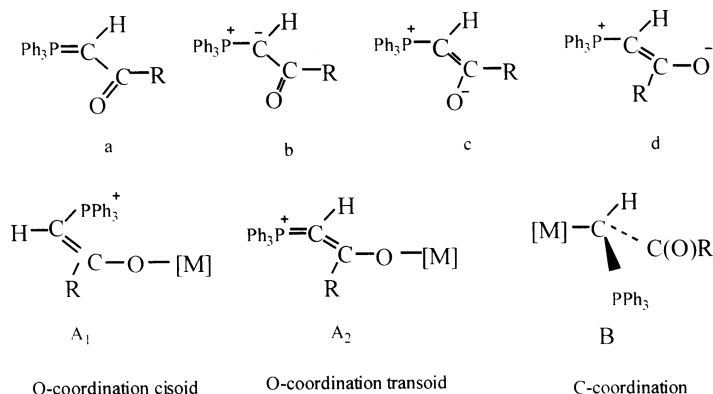
Among the phosphorus ylides of general stoichiometry R₃PCR'(R, R' = alkyl, alkoxy, etc.) the α -keto-stabilized ylides Ph₃PCHCOR have shown useful application in organometallic chemistry (due to their ambidentate character as ligands¹) and as reactants or valuable key intermediates in metal-mediated organic synthesis.^{1,2} This ambidentate character facilitates the preparation of stable metal complexes in which the ylide could be O-(A₁, A₂, Scheme 1)^{3–7} or C-coordinated (B).^{8–10} With both, modes rationalized in terms of the resonance forms a–c together with the isomeric formed. However, while a large number of compounds containing C-coordinated ylides are known, very few examples of O-bonded ones have been reported:^{3–7}

The phosphorus ylide complexes have been well investigated. They are versatile ligands for catalysts in a very small number of catalytic

Received December 2, 2004; accepted March 1, 2005.

We are grateful to the University of Bu-Ali-Sina for a grant and Mr. Zebarjadian for recording the NMR spectra.

Address correspondence to S. J. Sabounchei, Department of Chemistry, Bu-Ali-Sina University, Hamadan 65174, Iran. E-mail: jsabounchei@yahoo.co.uk

**SCHEME 1**

reactions, such as, for example, the hydrogenation of olefins and the cyclotrimerization and polymerization of acetylenes, but the most important application is in the industrially used SHOP process.¹¹ Thus complexes (I), (II), and (III) obtained from the reaction of the new α -carbonyl-stabilized ylide (I) with HgX_2 ($\text{X} = \text{Cl}, \text{Br}$ and I).

EXPERIMENTAL

Methanol (MeOH) and diethyl ether (Et_2O) were distilled over Mg or Na and CH_2Cl_2 over CaH_2 just before use. All other solvents were reagent grade and used without further purifications. Melting points were measured with a SMPI apparatus. Solid-state FT-IR spectra in the region of $400\text{--}4000\text{ cm}^{-1}$ using KBr pellets were obtained on a (Perkin Elmer) spectrophotometer. ^1H and ^{31}P spectra were obtained using a 90 MHz instrument at regional sophisticated instrumentation at Bu-Ali Sina University, and ^{13}C NMR spectra were measured with a BRUKER DRX-500 spectrometer. Elemental analyses were carried out at the Research Institute of Petroleum Industry.

Synthesis

Synthese of $\{(p\text{-tolyl})_3\text{PCHCOC}_6\text{H}_5\}$

Tri-*p*-tolylphosphine 0.152 g (0.5 mmol) was added in a chloroform (25 mL) to a solution of bromoacetophenone 0.099 g (0.5 mmol) in chloroform (20 mL) and the mixture was stirred for 4 h. The solution was filtered off and the precipitate was washed with diethylether and collected and dried. Then the reaction mixture was made basic by using NaOH (5%), and a white-yellow precipitate was obtained. Yield: 0.152 g

(70%). m.p. 188–190°C. Anal. found: C, 77.6; H, 6.1%; $C_{29}H_{27}OP \cdot 1.5 H_2O$ (MW = 449.5 gr). Calc.: C, 77.5; H, 6.68%.

Synthese of [$\{(p\text{-tolyl})_3PCHCOC_6H_5\} \cdot HgCl_2\}_2$]

A solution of 0.14 gr (0.5 mmol) of mercury (II) chloride in dry methanol (10 mL) was added to a methanolic solution of 0.211 gr (0.5 mmol) of ylide in dry methanol (10 mL). The white product formed by the slow evaporation of the solvent and dried in vacuum. Yield 0.347 gr (80.11%). m.p. 200–201°C. Anal. fund: C, 48.3; H, 4.1%; $C_{29}H_{27}OPHgCl_2 \cdot H_2O$ (MW = 711.59). Calc.: C, 48.91; H, 4.07%.

Synthese of [$\{(p\text{-tolyl})_3PCHCOC_6H_5\} \cdot HgBr_2\}_2$]

A solution of 0.211 gr (0.5 mmol) of the ylide in dry methanol (10 mL) was added to 0.18 gr of mercury (II) bromide in dry methanol (10 mL) and the mixture was stirred for 4 h. The solvent was then removed in vacuo. The white product obtained was washed with ice-cold methanol and dried in vacuum. Yield 0.294 gr (75.15%). m.p. 193–195°C. Anal. found: C, 43.5; H, 3.7%; $C_{29}H_{27}OPHgBr_2 \cdot H_2O$ (MW = 800.39). Calc.: C, 43.48 H, 3.62%.

Synthese of [$\{(p\text{-tolyl})_3PCHCOC_6H_5\} \cdot HgI_2\}_2$]

A solution of 0.211 gr (0.5 mmol) of the ylide in methanol (10 mL) was added to a solution of 0.227 gr (0.5 mmol) of mercury (II) iodide in methanol (10 mL). On concentration by removing the solvent in vacuum, a pale yellow, light-sensitive crystalline solid was obtained. The crystals were washed with benzene and dried in vacuo. Yield 0.307 gr (70.2%). m.p. 200–202°C. Anal. found: C, 37.3; H, 3.4%; $C_{29}H_{27}OPHgI_2 \cdot 2H_2O$ (MW = 912.39). Calc.: C, 38.14; H, 3.39%.

TABLE I ν (CO) of Selected Compound (I) and Their Complexes With Mercury (II) Halides

Compound	ν (CO) cm^{-1}	Ref.
$Ph_3PCHCOCH_3$ (APPY)	1530	12
$Ph_3PCHCOPh$ (BPPY)	1525	12
Ph_3PCHCO (p-tolyl) ₃	1528	This article
C-coordination		
BPPY. $HgCl_2$	1635	12
BPPY. $HgBr_2$	1630	12
$[Ph_3PCHCO(p\text{-tolyl})_3HgCl_2]_2$	1597	This article
$[Ph_3PCHCO(p\text{-tolyl})_3HgBr_2]_2$	1633	This article
$[Ph_3PCHCO(p\text{-tolyl})_3HgI_2]_2$	1597	This article

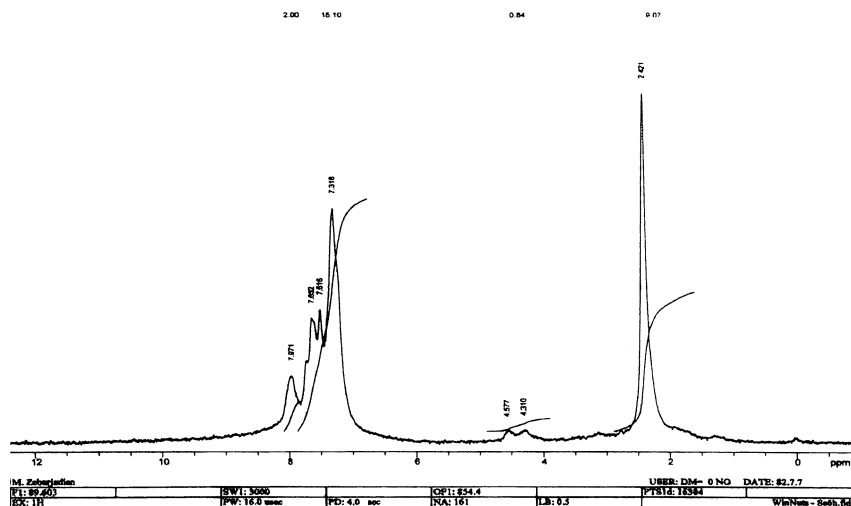


FIGURE 1 ^1H NMR $\{(\text{p-tolyl})_3\text{PCHOC}_6\text{H}_5\}$ in CDCl_3 at 25°C .

RESULTS AND DISCUSSION

The ν (CO), which is sensitive to complexation, occurs at 1528 cm^{-1} in the parent ylide, as in the case of other resonance stabilized ylides. Coordination of the ylide through carbon causes an increase in ν (CO) while for O-coordination a lowering of ν (CO) is expected. The IR absorption

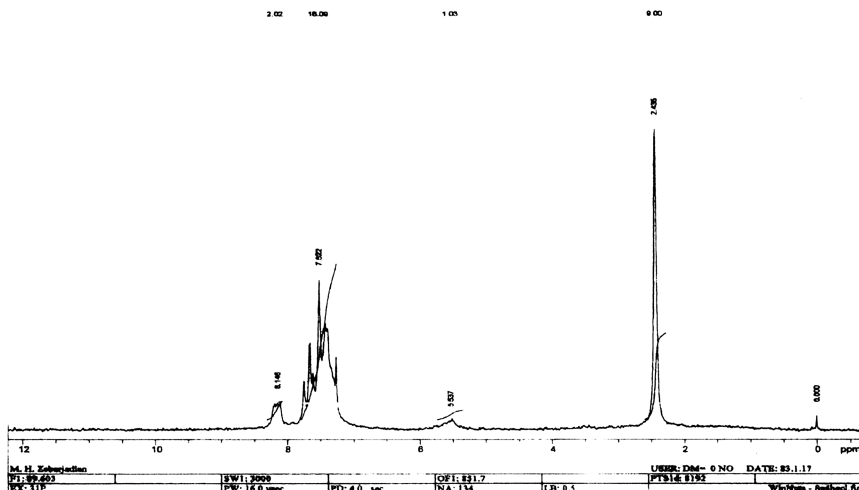


FIGURE 2 ^1H NMR $\{(\text{p-tolyl})_3\text{PCHOC}_6\text{H}_5\cdot\text{HgCl}_2\}_2$ in CDCl_3 at 25°C .

TABLE II ^1H and ^{31}P NMR Data of (I) and Its Complexes With Mercury(II) Halides. s-singlet; d-doublet; m-multiplet; br-broad

Compound	^1H NMR			^{31}P NMR
	$\delta(\text{CH})$	$^2J_{(\text{P}-\text{H})}$	δPh	
$\text{Ph}_3\text{PCHCO}(\text{p-tolyl})_3$	4.4(d)	24.39	7.3–7.95(m)	12.98(s)
$[\text{Ph}_3\text{PCHCO}(\text{p-tolyl})_3.\text{HgCl}_2]_2$	5.54(br)	—	7.26–8.13(m)	24.4(s)
$[\text{Ph}_3\text{PCHCO}(\text{p-tolyl})_3.\text{HgBr}_2]_2$	5.2(br)	—	7.25–8.01(m)	22.22(s)
$[\text{Ph}_3\text{PCHCO}(\text{p-tolyl})_3.\text{HgI}_2]_2$	5.29(br)	—	7.52–8.06(m)	18.5(s)

In CDCl_3 , 90 MHz. values (ppm) are relative to internal TMS and external 85% phosphoric acid.

bands observed for the three complexes at 1597, 1633, and 1597 cm^{-1} indicate coordination of the ylide through carbon. The $\nu(\text{P}^+-\text{C}^-)$, which is also diagnostic of the coordination, occurs at 879 cm^{-1} in $(\text{p-tolyl})_3\text{P}^+-\text{CH}$ (Table I). These assignments were confirmed by comparing the IR spectra of the corresponding ^{13}C substituted ylides.¹² In the present study, the $\nu(\text{P}^+-\text{C}^-)$ values for all three complexes were shifted to lower frequencies and observed at 805 cm^{-1} for three complexes, suggesting some removal of electron density in the P–C bond.

In the ^1H NMR spectra for three complexes, the CH ylide proton shifted downfield compared to that of the free ylide (Figures 1 and 2) as a consequence of the inductive effect of the metal center. The expected

TABLE III ^{13}C NMR Data of Benzoylmethelenparatolylephosphoran and Corresponding Complexes With Mercury(II) Halides. s-singlet; d-double; o-ortho; m-meta; p-para; i-ipso carbon. Recorded in CDCl_3

Compound	$\text{Ph}_3\text{PCHCO}(\text{p-tolyl})_3$	Complex (I)	Complex (II)	Complex (III)
3CH_3	22.2(s)	22.2(s)	22.18(s)	22.22(s)
CH	51.8(d)	—	—	44.15(s)
$^2J_{(\text{p}-\text{c})}$	113			
$\text{CO}-\text{Ph}(\text{o})$	127.38(s)	119.3(s)	119.89(s)	117.87(s)
$\text{CO}-\text{Ph}(\text{m})$	128.13(s)	119.87(s)	120.62(s)	118.6(s)
$\text{CO}-\text{Ph}(\text{p})$	133.63(s)	133.91(s)	132.53(s)	134.88(s)
$\text{CO}-\text{Ph}(\text{i})$	142.58(s)	145.48(s)	145.13(s)	146.34(s)
$\text{P}-(\text{p-tolyl})_3(\text{o})$	132.5(d)	133.99(d)	134.02(d)	133.83(d)
$^2J_{(\text{p}-\text{c})}$	10.6	10.5	10.5	10.6
$\text{P}-(\text{p-tolyl})_3(\text{m})$	129.57(d)	131.1(d)	130.95(d)	131.38(d)
$^3J_{(\text{p}-\text{c})}$	12.25	13	13	13
$\text{P}-(\text{p-tolyl})_3(\text{p})$	130.67(s)	129.48(s)	133.18(s)	129.56(s)
$\text{P}-(\text{p-tolyl})_3(\text{i})$	124.49(d)	129.21(s)	128.95(s)	128.92(s)
$^1J_{(\text{p}-\text{c})}$	95.6			
CO	184.93(s)	193.93(s)	192.41(s)	192.89

downfield shifts of the ^{31}P and ^1H singles for the PCH group upon complexation were observed in their corresponding spectra (Figures 1 and 2, Table II). The appearance of single signals for the PCH group in the ^{31}P and ^1H NMR spectra indicates the presence of only one molecule for all three complexes, as expected for C-coordination.

The ^{13}C NMR data of the complexes and the title ylide are listed in Table III along with possible assignments. The ^{13}C NMR shifts of the CO group in the complexes are around 190 ppm, lower than 184 ppm noted for the same carbon in the parent ylide, indicating a much lower shielding of carbon of the CO group in the complexes.

REFERENCES

- [1] A. W. Johnson, W. C. Kaska, K. A. O. Starzewski, and D. A. Dixon, *Ylides and Imines of Phosphorus* (John Wiley and Sons, New York, 1993), Chap. 14 and references given therein.
- [2] M. E. Jung and S. A. Abrechet, *J. Org. Chem.*, **53**, 423 (1988).
- [3] J. Buckle, P. G. Harrison, T. J. King, and J. A. Richardes, *J. Chem. Soc. Chem. Commun.*, 1104 (1972).
- [4] J. Buckle and P. G. Harrison, *J. Organomet. Chem.*, **49**, C17 (1973).
- [5] I. Kawafune and G. Matsubayashi, *Inorg. Chem. Acta*, **70**, 1 (1983).
- [6] R. Uson, J. Fornies, R. Navarro, P. Espinet, and C. Mendivil, *J. Organomet. Chem.*, **290**, 125 (1985).
- [7] J. A. Albanese, D. A. Staley, A. L. Rheingold, and J. L. Burmeister, *Inorg. Chem.*, **29**, 2209 (1990).
- [8] S. J. Sabounchei and K. Karami, *Phosphorus, Sulfur, and Silicon*, **178**, 1559 (2003).
- [9] H. Koezuka, G. Matsubayashi, and T. Tanaka, *Inorg. Chem.*, **15**, 417 (1976).
- [10] G. Fronza, P. Bravo, and C. Ticozzi, *J. Organomet. Chem.*, **157**, 299 (1978).
- [11] (a) K. W. Baure, R. S. Chung, and H. C. Glockner, P. U. S. Patent 3647914, (1969);
(b) K. W. Mason, and R. F. Glockner, P. U. S. Patent 3647914, (1972).
- [12] M. Kalyanasundari, K. T. Panchanatheswaran, and R. Huowen, *J. Organomet. Chem.*, **491**, 403 (1995).